

1-*tert*-Butyl-3-(trifluoromethyl)-1*H*-pyrazol-5-yl
4-chlorobenzoateYang Li,^{a*} Jiang-Sheng Li,^a
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Key indicators

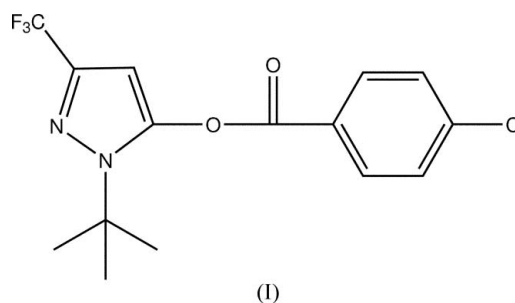
Single-crystal X-ray study
T = 294 K
Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$
Disorder in main residue
R factor = 0.038
wR factor = 0.096
Data-to-parameter ratio = 12.1For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.In the title compound, $\text{C}_{15}\text{H}_{14}\text{ClF}_3\text{N}_2\text{O}_2$, the benzene and
pyrazolyl rings form a dihedral angle of $15.9 (3)^\circ$.

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Comment

In order to explore the Fries-type rearrangement of 5-
pyrazolyl esters, the products of which are active ingredients
of herbicides (Konotsune & Kawakubo, 1975; Tanaka *et al.*,
1999; Mayer *et al.*, 2000), the title compound, (I), was prepared
via 4-chlorobenzoylation of 5-hydroxylpyrazole. The molec-
ular structure of (I) (Fig. 1 and Table 1) shows that the
benzene and pyrazolyl rings are essentially planar, with mean
deviations of 0.005 and 0.003 \AA , respectively. The dihedral
angle formed between these rings is $15.9 (3)^\circ$.

Experimental

To a mixture of benzene (20 ml) and water (2 ml) were added 1-*tert*-
butyl-3-(trifluoromethyl)pyrazolone (Liu & Li, 2004; 0.83 g, 4 mmol),
anhydrous sodium carbonate (0.28 g, 2 mmol) and a catalytic amount
of tetrabutylammonium bromide. 4-Chlorobenzoyl chloride (0.70 g,
4 mmol) in benzene (5 ml) was added dropwise within 30 min at
283 K, and stirred at room temperature for 1 h. The aqueous layer
was removed and the benzene layer was evaporated under reduced
pressure. The crude product was purified by silica-gel column chro-
matography (ethyl acetate/petroleum ether = 1:25) to afford 1.14 g of
(I) as a white solid in 82% yield (m.p. 357–359 K). ¹H NMR (CD_3Cl):
 δ 1.67 (s, 9H), 6.53 (s, 1H), 7.55 (d, 2H, *J* = 9 Hz), 8.11 (d, 2H, *J* =
9 Hz). ¹³C NMR (CDCl_3): δ 160.8, 144.4, 141.4, 139.0, 131.6 (2C),
129.6 (2C), 126.1, 122.1, 95.1, 61.1, 29.2 (3C). Suitable crystals were
obtained by the slow evaporation of a mixture of ethyl acetate and *n*-
hexane.

Crystal data

 $\text{C}_{15}\text{H}_{14}\text{ClF}_3\text{N}_2\text{O}_2$
M_r = 346.73
Monoclinic, $P2_1/c$
a = 7.270 (2) \AA
b = 11.927 (4) \AA
c = 18.528 (6) \AA
 β = 98.240 (5) $^\circ$
V = 1589.9 (8) \AA^3
Z = 4*D_x* = 1.449 Mg m^{-3}
Mo *K* α radiation
Cell parameters from 2353
reflections
 θ = 2.2–24.1 $^\circ$
 μ = 0.28 mm^{-1}
T = 294 (2) K
Block, colourless
0.34 × 0.29 × 0.20 mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.892$, $T_{\max} = 0.945$
 8641 measured reflections

3242 independent reflections
 1832 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$
 $\theta_{\text{max}} = 26.4^\circ$
 $h = -9 \rightarrow 8$
 $k = -14 \rightarrow 10$
 $l = -23 \rightarrow 23$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.096$
 $S = 1.00$
 3242 reflections
 268 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0291P)^2 + 0.5765P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.002$
 $\Delta\rho_{\text{max}} = 0.18 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.18 \text{ e } \text{\AA}^{-3}$

Table 1

Selected geometric parameters (\AA , $^\circ$).

N1—N2	1.349 (2)	C8—O1	1.369 (2)
C1—N1	1.493 (3)	C8—N1	1.349 (2)
C1—C3	1.520 (3)	C9—O1	1.365 (2)
C5—N2	1.316 (3)	C9—O2	1.185 (3)
C5—C7	1.387 (3)	C13—C11	1.733 (2)
C7—C8	1.354 (3)		
C8—O1—C9	120.13 (17)	N1—N2—C5	104.29 (16)
N2—N1—C1	120.41 (16)	N2—C5—C7	113.5 (2)
N2—N1—C8	110.28 (16)	C5—C7—C8	102.78 (19)
C1—N1—C8	129.31 (17)		
O1—C9—C10—C11	-168.0 (2)	C3—C1—N1—C8	-62.1 (3)
O2—C9—O1—C8	-3.0 (3)	C7—C8—O1—C9	4.9 (4)

H atoms were placed in calculated positions, with C—H distances of 0.93 (aromatic) and 0.96 \AA (CH_3). They were included in the refinement in the riding-model approximation with isotropic displacement parameters set equal to $1.2U_{\text{eq}}$ of the carrier atom for the aromatic H atoms, and $1.5U_{\text{eq}}$ of the carrier atom for CH_3 H atoms. The CF_3 group was found to be disordered and three distinct conformations were observed. The F atom was refined anisotropic

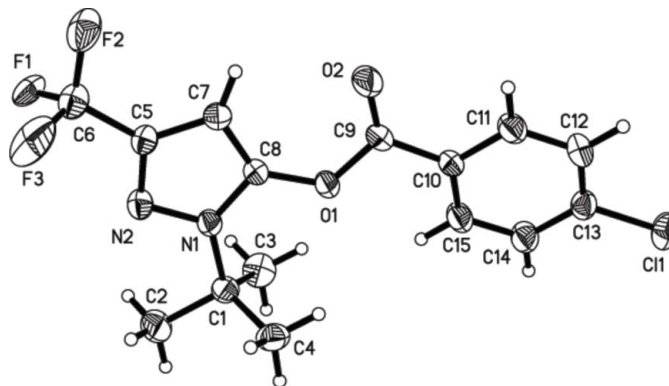


Figure 1

A view of (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii. Only the major component of the disordered CF_3 group is shown.

and the site occupancies were fixed at 0.5:0.4:0.1 at the final refinement stage.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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